

Name: _____
Hour: _____ Date: _____

Chemistry: *Ions in Chemical Formulas*

Write the correct formula for the compound formed by each of the following pairs of ions.

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|---|-----------|
| 1. Na^{1+} F^{1-} | 1. _____ |
| 2. K^{1+} S^{2-} | 2. _____ |
| 3. Ni^{2+} SO_4^{2-} | 3. _____ |
| 4. Al^{3+} O^{2-} | 4. _____ |
| 5. Ca^{2+} ClO_3^{1-} | 5. _____ |
| 6. NH_4^{1+} P^{3-} | 6. _____ |
| 7. Cu^{1+} NO_3^{1-} | 7. _____ |
| 8. Cu^{2+} NO_3^{1-} | 8. _____ |
| 9. Pb^{4+} O^{2-} | 9. _____ |
| 10. Li^{1+} CO_3^{2-} | 10. _____ |

For each of the following compounds, write...

A) the symbols of the ions in the compound, and

B) the number of each ion in one molecule of that compound.

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| 11. CaI_2 | 11. _____ |
| 12. Na_2CO_3 | 12. _____ |
| 13. $\text{Ga}(\text{ClO}_3)_3$ | 13. _____ |
| 14. CuF_2 | 14. _____ |
| 15. $(\text{NH}_4)_3\text{PO}_4$ | 15. _____ |
| 16. FeSO_4 | 16. _____ |
| 17. $\text{Mg}(\text{NO}_3)_2$ | 17. _____ |
| 18. NH_4NO_2 | 18. _____ |
| 19. $\text{KC}_2\text{H}_3\text{O}_2$ | 19. _____ |
| 20. $\text{Na}_2\text{Cr}_2\text{O}_7$ | 20. _____ |