Chemistry: Ionic Compounds: Traditional System of Nomenclature

Write the traditional name of each of the following compounds.

1. CuCl
2. FeO
3. PbO₂
4. Sn(SO₄)₂
5. Au₃N
6. Fe₂S₃
7. Cu₃(PO₄)₂
8. Sn(ClO₃)₂
9. AuF
10. PbCr₂O₇

Write the chemical formula for each of the given traditional names.

11. stannic carbonate
12. cuprous oxalate
13. plumbic bicarbonate
14. aurous phosphide
15. ferric hydroxide
16. cupric nitride
17. auric iodate
18. plumbous sulfite
19. ferrous chloride
20. stannous cyanide