# Name: \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Hour: \_\_\_\_ Date: \_\_\_\_\_\_\_\_\_\_\_

# Chemistry: *Gas Laws with One Term Constant*

Solve the following problems. Show your work and include correct units for full credit.

1. A gas has an initial volume of 15 L. If the temperature increases from 330 K to 450 K, find the new volume.

2. A gas exerts 1.2 atm of pressure. If the temperature is raised from 25oC to 100oC, find the new pressure.

3. A sample of oxygen takes up 34 dm3 of space when it is under 500 kPa of pressure. When the pressure is changed to 340 kPa, find the new volume.

4. The pressure of some N2 drops from 315 kPa to 220 kPa. If the initial volume is 1.4 L, find the new volume.

5. The pressure of neon changes from 786 mm Hg to 1811 mm Hg. If the initial temperature 87oC, what is the new temperature (in oC)?

6. When the temperature of a gas changes, its volume decreases from 12 cm3 to 7 cm3. If the **final** temperature is measured to be 18oC, what was the initial temperature (in oC)?

Answers: 1. 20.5 L 2. 1.5 atm 3. 50 dm3 4. 2.0 L 5. 556oC 6. 226oC